

TOTAL SYNTHESIS OF L-2,2-DIFLUORO-2-DEOXY-MYO-INOSITOL 1,4,5-TRISPHOSPHATE, A POTENT INHIBITOR OF THE ENZYMES OF D-MYO-INOSITOL 1,4,5-TRISPHOSPHATE METABOLISM

Deborah A Sawyer[#] and Barry V L Potter^{##}

[#]Department of Chemistry, University of Leicester, Leicester LE1 7RH, UK

and

^{*}School of Pharmacy and Pharmacology, University of Bath, Claverton Down, Bath BA2 7AY, UK

(Received 24 September 1991)

Abstract: The synthesis of the enantiomers of 2,2-difluoro-2-deoxy-*myo*-inositol 1,4,5-trisphosphate is reported. L-2,2-difluoro-2-deoxy-*myo*-inositol 1,4,5-trisphosphate is a potent inhibitor of 3-kinase and 5-phosphatase.

It is now generally accepted that the cyclitol polyphosphate, D-*myo*-inositol 1,4,5-trisphosphate Ins(1,4,5)P₃ [(1), Fig. 1], is a second messenger which releases Ca²⁺ from intracellular stores^{1,2} via a receptor which has now been isolated³, cloned and sequenced^{4,5} and which, when reconstituted, is gated in response to Ins(1,4,5)P₃⁶. A major challenge is now the elucidation of the structural basis for interaction of Ins(1,4,5)P₃ both with its receptor and with the metabolic enzymes, Ins(1,4,5)P₃ 3-kinase and 5-phosphatase. This will facilitate the rational design of agonists, antagonists and enzyme inhibitors which may be of therapeutic value. Recent progress in inositol phosphate chemistry^{7,8} and the molecular recognition of inositol phosphates by proteins has been reviewed⁹.

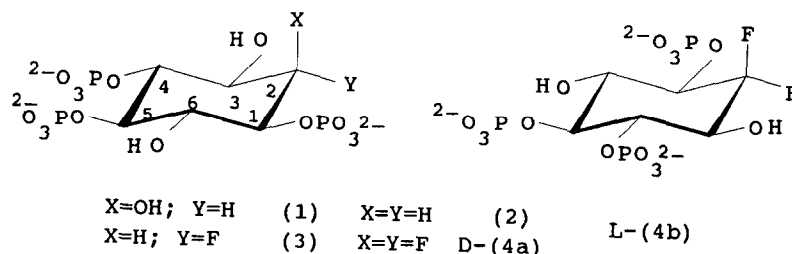


Fig. 1: Structures of D-Ins(1,4,5)P₃ (1), related synthetic analogues and the fluorinated D- and L- enantiomers of 2,2-difluoro-2-deoxy-*myo*-inositol 1,4,5-trisphosphate (4a) and (4b).

Ring-modified and phosphate-modified inositol analogues have been synthesized^{7,8} in attempts to understand the role of three phosphate and hydroxyl groups of Ins(1,4,5)P₃ in receptor binding specificity and stimulation. Isosteric replacement of a hydroxyl group with fluorine¹⁰ has led to fluorinated inositol analogues¹¹⁻¹⁴ and inositol polyphosphate analogues¹⁵⁻¹⁹. D-3-fluoro-3-deoxy-*myo*-inositol was found to inhibit cell growth in NIH 3T3 cells¹¹ and 5-fluoro-5-deoxy-*myo*-inositol is taken up by L1210 cells and incorporated into cellular phospholipid by phosphatidylinositol synthase²⁰, although 5,5-difluoro-5-deoxy-*myo*-inositol is a much poorer substrate¹³. Reports of biological activity for non-fluorinated ring-modified analogues of Ins(1,4,5)P₃, including 2-deoxy-Ins(1,4,5)P₃ [(2), Fig. 1), have appeared²¹⁻²⁶. We recently reported the first biological evaluation of some fluorinated inositol phosphate analogues¹⁸ and studied the interaction of the two synthetic analogues DL-2-deoxy-2-fluoro-*scyllo*-inositol 1,4,5-trisphosphate [DL-2-F-Ins(1,4,5)P₃] (3) and DL-2,2-difluoro-2-deoxy-*myo*-inositol 1,4,5-trisphosphate [DL-2,2-F₂-Ins(1,4,5)P₃] DL-(4ab) with the Ca²⁺-releasing Ins(1,4,5)P₃ receptor and the metabolic enzymes Ins(1,4,5)P₃ 5-phosphatase and 3-kinase. Racemic 2,2-F₂-Ins(1,4,5)P₃, although in other respects an excellent Ins(1,4,5)P₃ analogue, was apparently and unexpectedly not a substrate for 5-phosphatase. Since 2-deoxy-Ins(1,4,5)P₃ (2) is known to be a substrate for this enzyme²², we suspected that L-2,2-difluoro-2-deoxy-*myo*-inositol trisphosphate L-(4b) might be acting as a potent 5-phosphatase inhibitor. We have now resolved (4ab) into individual D- and L- enantiomers. We report here the total synthesis of these enantiomers.

DL-2,2-difluoro-2-deoxy-*myo*-inositol 1,4,5-trisphosphate (4ab) was synthesized by diethylaminosulphur trifluoride fluorination of 1-*O*-allyl-3,6-di-*O*-benzyl-4,5-*O*-isopropylidene-*myo*-2-inosose²⁸ (6) [prepared from (5)] to give (7), removal of non-benzylic protecting groups to give the racemic triol (9ab); followed by bis(2-cyanoethyl)-N,N-diisopropylaminophosphine phosphitylation, oxidation of the resulting trisphosphite with *t*-butyl hydroperoxide and deprotection using sodium in liquid ammonia, as described for Ins(1,4,5)P₃²⁹. Racemic (4ab) was purified by ion-exchange chromatography. The individual enantiomers were synthesized by preparation of the diastereoisomeric mixed 1-*O*-[S(-)-camphanyl] esters (11) and (12) of DL-2,2-difluoro-2-deoxy-3,6-di-*O*-benzyl-4,5-*O*-isopropylidene-*myo*-inositol (10ab) followed by their separation by flash chromatography on silica gel (Fig. 2). The absolute configuration of the diastereoisomeric camphanate of the D-enantiomer, D-(4a), was determined by single crystal X-ray crystallography³⁰ (Fig. 3) and, after removal of the chiral auxiliary and the isopropylidene group, the pure enantiomers of 2,2-difluoro-3,6-dibenzyl-*myo*-inositol (9a) and (9b) were individually polyphosphorylated, deblocked and purified as for the racemic modification to give D-(4a) and L-(4b) respectively. The enantiomers were purified on an ion-exchange column of DEAE Sephadex A-25 using a gradient of triethylammonium bicarbonate buffers, pH 8.0,

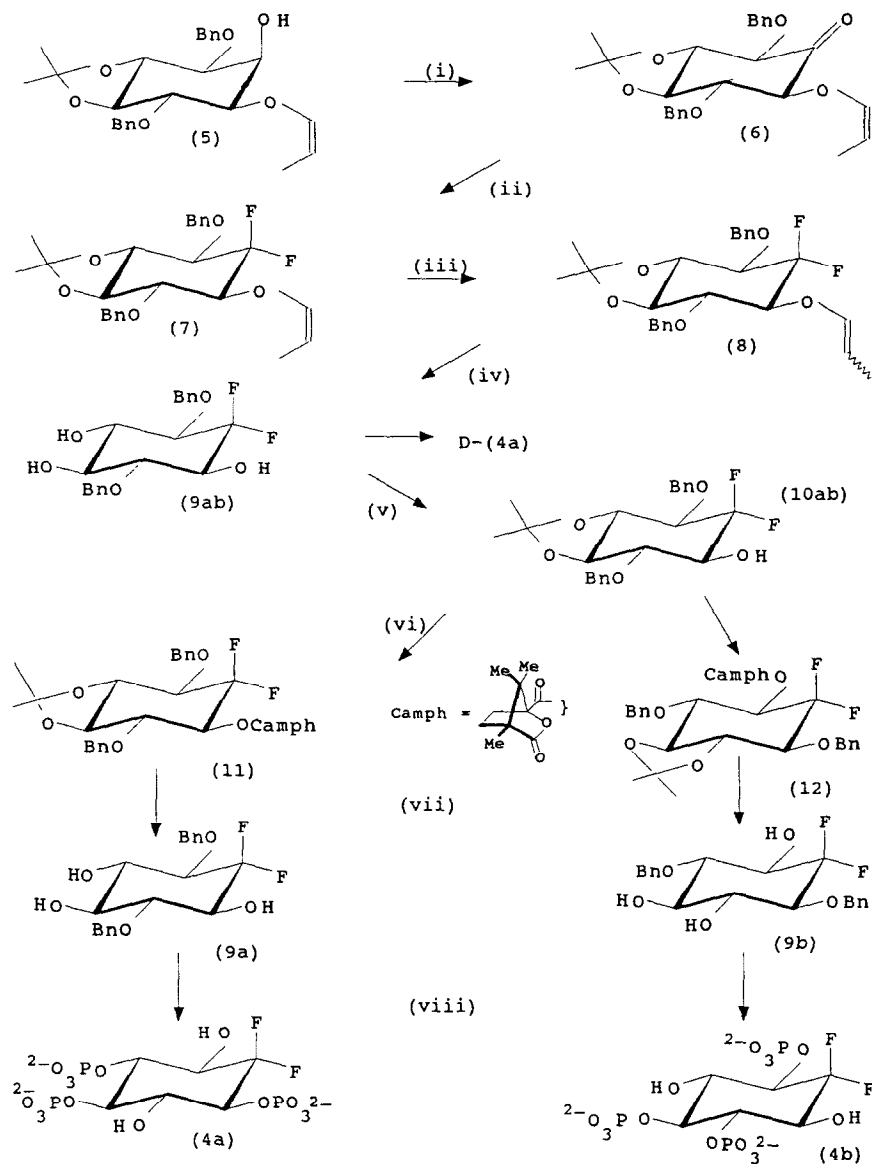


Fig. 2: Synthesis of fluorinated inositol phosphate analogues and resolution of triol (9ab) via the diastereomeric camphanates (11) and (12)

Reagents and conditions: i (CH₃CO)₂O (21 equiv)/DMSO, RT 16h, (b) aq NaHCO₃, ii(a) DAST (4 equiv)/CH₂Cl₂, RT 5h, (b) aq NaHCO₃, iii(a) DABCO (0.2 equiv)/(Ph₃P)₃RhCl (0.08 equiv)/EtOH:H₂O (9:1), reflux 2h, (b) H₂O, iv(a) M HCl-MeOH (1:5), reflux 30 min, (b) excess solid NaHCO₃, v(a) (MeO)₂CMe₂ (40 equiv), *p*-TsOH/Me₂CO, RT 2h, (b) Et₃N, solid NaHCO₃; vi(a) *S*-(-)-camphanic acid chloride (2 equiv)/C₅H₅N, RT 12h, (b) H₂O; vii(a) NaOH/MeOH, reflux 1h, (b) M HCl-methanol (1:5), reflux 30 min (c) excess solid NaHCO₃; viii(a) (NCCH₂CH₂O)₂P-NPr₂ (6 equiv), tetrazole (10 equiv)/CH₂Cl₂, RT 1h, (b) 70% aq BuOOH, -78°C 1h, (c) Na-liq NH₃, -78°C 15 min, (d) H⁺-Dowex.

and were isolated as their triethylammonium salts and assayed by quantitative phosphate analysis.

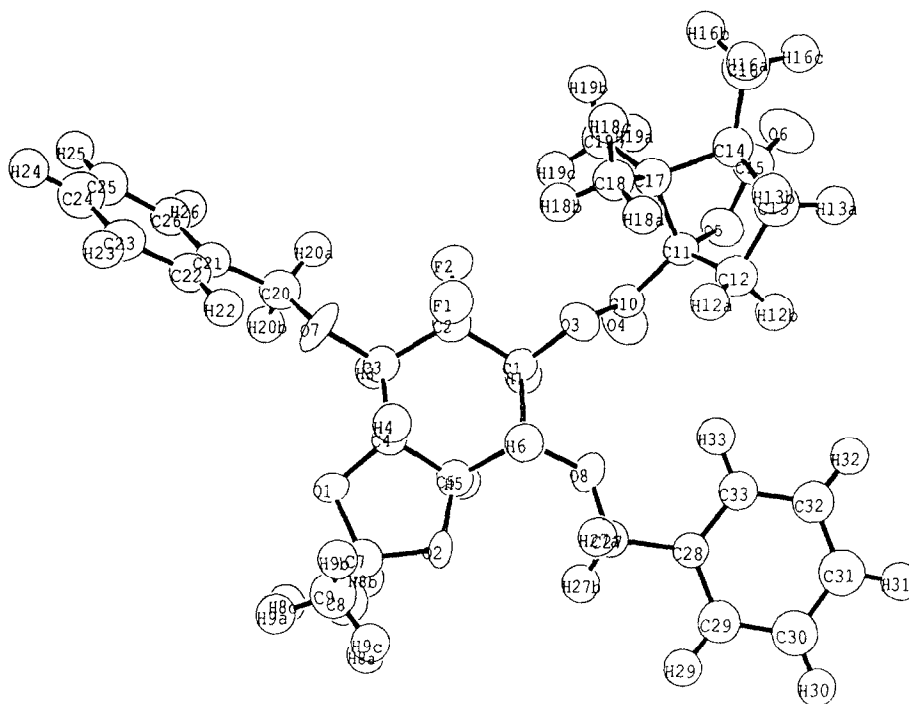


Fig. 3: Absolute configuration of 1-*O*-*S*-(-)-camphanyl-*D*-2,2-difluoro-2-deoxy-3,6-di-*O*-benzyl-4,5-*O*-isopropylidene-*myo*-inositol (11) determined by X-ray analysis.

Using fluorinated Ins(1,4,5) P_3 analogues we attempted to probe the role of the unique axial 2-hydroxyl group in determining the affinity and specificity of Ins(1,4,5) P_3 for its receptor and metabolic enzymes¹⁸. We previously determined the EC_{50} of DL-(4ab) to be $0.41\mu M$ ¹⁸ [EC_{50} for D-Ins(1,4,5) P_3 , $0.13\mu M$] D-(4a) and L-(4b) had EC_{50} values of $0.21\mu M$ and $53\mu M$ respectively. D-(4a) was an excellent analogue of Ins(1,4,5) P_3 , being a full agonist in Ca^{2+} release, a substrate for erythrocyte membrane 5-phosphatase and rat brain 3-kinase, and bound with high affinity by the enzymes. By contrast, L-(4b) did not release Ca^{2+} effectively and was not a substrate for 5-phosphatase, but was a relatively potent inhibitor ($K_i = 26\mu M$). Moreover, it inhibited the phosphorylation of Ins(1,4,5) P_3 by 3-kinase potently ($K_i = 11.5\mu M$) [NB K_i for L-Ins(1,4,5) $P_3 = 292\mu M$ ²¹].

Modification of the 2-position of L-Ins(1,4,5)P₃ by fluorination has led to the production of a moderately potent 5-phosphatase inhibitor and, most interestingly, of the first small molecule 3-kinase inhibitor. This inhibitor has the advantage that it is a very poor agonist for Ca²⁺ release. Although the precise reasons for these inhibitory activities are not yet known, it is clear that L-(4b) represents a novel lead in the design of further molecular entities for pharmacological intervention in the polyphosphoinositide signalling pathway.

ACKNOWLEDGEMENTS: This work was supported by SERC (Molecular Recognition Initiative). We thank the University of Leicester for a Research Scholarship (to DS), S Safrany and S R Nahorski for biological evaluations, Dr J Fawcett for X-ray crystallographic studies on synthetic precursors and S Alston for manuscript preparation. BVLP is a Lister Institute Fellow.

REFERENCES

1. Berridge, M.J. *Annu. Rev. Biochem.* **1987**, *56*, 159.
2. Berridge, M.J.; Irvine, R.F. *Nature* **1989**, *341*, 197.
3. Supattapone, S; Worley, P.F.; Baraban, J.M.; Snyder, S.H. *J. Biol. Chem.* **1988**, *263*, 1530.
4. Furuichi, T.; Yoshikawa, S.; Miyawaki, A.; Wada, K.; Maeda, N.; Mikoshiba, K. *Nature (Lond.)* **1989**, *342*, 32.
5. Mignery, G.A.; Newton, C.L.; Archer, B.T.; Südhof, T.C. *J. Biol. Chem.* **1990**, *265*, 12679.
6. Ferris, C.D.; Huganir, R.L.; Supattapone, S.; Snyder, S.H. *Nature* **1989**, *342*, 87.
7. Billington, D.C. *Chem. Soc. Rev.* **1989**, *18*, 83.
8. Potter, B.V.L. *Nat. Prod. Reports* **1990**, *7*, 1.
9. Nahorski, S.R.; Potter, B.V.L. *Trends Pharmacol. Sci.* **1989**, *10*, 139.
10. Schlosser, M. *Tetrahedron* **1978**, *34*, 3.
11. Kozikowski, A.P.; Fauq, A.H.; Powis, G.; Melder, D.C. *J. Amer. Chem. Soc.* **1990**, *112*, 4528.
12. Jiang, C.; Moyer, J.D.; Baker, D.C. *J. Carbohydr. Chem.* **1987**, *6*, 319.
13. Jiang, C.; Schedler, D.J.A.; Morris, P.E., Jr.; Zayed, A.-H.A.; Baker, D.C. *Carb. Res.* **1990**, *207*, 277.
14. Offer, J.L.; Metcalfe, J.C.; Smith, G.A. *J. Chem. Soc. Chem. Commun.* **1990**, 1312.
15. Marecek, J.F.; Prestwich, G.D. *Tet. Lett.* **1989**, *30*, 5401.

16. Boehm, M.F.; Prestwich, G.D. *Tet. Lett.* **1988**, *29*, 5217.
17. Ley, S.V.; Parra, M.; Redgrave, A.J.; Sternfeld, F.; Vidal, A. *Tet. Lett.* **1989**, *30*, 3557.
18. Safrany, S.T.; Sawyer, D.; Wojcikiewicz, R.J.H.; Nahorski, S.R.; Potter, B.V.L. *FEBS Lett.* **1990**, *276*, 91.
19. Kozikowski, A.P.; Fauq, A.H.; Aksoy, I.A.; Seewald, M.J.; Powis, G. *J. Amer. Chem. Soc.* **1990**, *112*, 7403.
20. Moyer, J.D.; Reizes, O.; Surender, A.; Jiang, C.; Malinowski, N.; Baker, D.C. *Mol. Pharmacol.* **1988**, *33*, 683.
21. Polokoff, M.A.; Bencen, G.H.; Vacca, J.P.; deSolms, S.J.; Young, S.D.; Huff, J.R. *J. Biol. Chem.* **1988**, *263*, 11922.
22. Hirata, M.; Watanabe, Y.; Ishimatsu, T.; Ikebe, T.; Kimura, Y.; Yamaguchi, K.; Ozaki, S.; Koga, T. *J. Biol. Chem.* **1989**, *264*, 20303.
23. Hirata, M.; Yanaga, F.; Koga, T.; Ogasawara, T.; Watanabe, Y.; Ozaki, S. *J. Biol. Chem.* **1990**, *265*, 8404.
24. Safrany, S.T.; Wojcikiewicz, R.J.H.; Strupish, J.; Dubreuil, D.; Cleophax, J.; Gero, S.D.; Nahorski, S.R.; Potter, B.V.L. *FEBS Lett.* **1991**, *278*, 252.
25. Seewald, M.J.; Aksoy, I.A.; Powis, G.; Fauq, A.H.; Kozikowski, A.P. *J. Chem. Soc. Chem. Commun.* **1990**, 1638.
26. Liu, C.; Nahorski, S.R.; Potter, B.V.L. *J. Chem. Soc. Chem. Commun.* **1991**, 1014.
27. Denis, G.V.; Ballou, C.E. *Cell Calcium* **1991**, *12*, 395.
28. Gigg, J.; Gigg, R.; Payne, S.; Conant, R. *J. Chem. Soc. Perkin Trans I* **1987**, 1757; 2411.
29. Cooke, A.M.; Gigg, R.; Potter, B.V.L. *Tet. Lett.* **1987**, *28*, 2305.
30. Crystal data for $C_{33}H_{38}O_8F_2$, $M = 600.65$, Monoclinic, space group = $P2_1$, $a = 8.602(10)$, $b = 9.090(3)$, $c = 19.650(23)$ Å, $\beta = 95.24(5)^\circ$, $U = 1530(3)$ Å³, $Z = 2$, $\mu = 0.62$ cm⁻¹, (Mo-K α) = 0.7107 Å, $F(000) = 636.0$, $D_c = 1.30$ g cm⁻³; unique data = 2589, observed data = 1142, $I > 3\sigma(I)$. The structure was solved using the TREF option of SHELXS86 and the program SHELX-76. Carbon was refined isotropically; oxygen and fluorine anisotropically; hydrogen atoms were included in calculated positions (C-H = 1.08 Å). Refinement converged at $R = 0.069$ and $R_w = 0.068$.